

Nickel-Catalyzed Allylic C(sp³)–F Bond Activation of Trifluoromethyl Groups via β -Fluorine Elimination: Synthesis of Difluoro-1,4-dienes

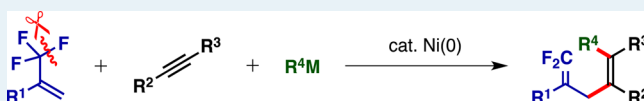
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Supporting Information

ABSTRACT: The nickel-catalyzed defluorinative coupling of 2-trifluoromethyl-1-alkenes and alkynes with the aid of Et₃SiH provides 1,1-difluoro-1,4-dienes under mild reaction conditions. This reaction involves selective allylic C(sp³)–F bond activation via β -fluorine elimination from nickelacyclopentenes.

KEYWORDS: C–F bond activation, nickel catalysis, trifluoromethylalkenes, alkynes, 1,4-dienes, β -fluorine elimination

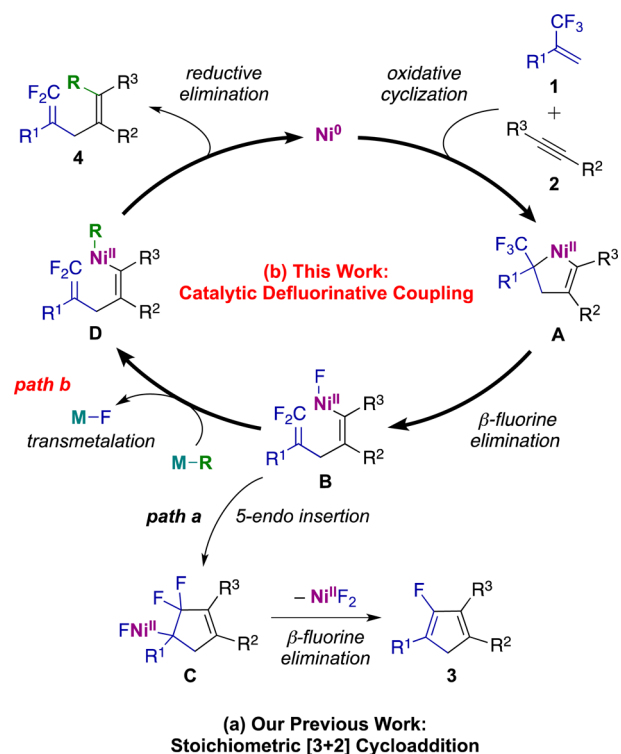


Carbon–fluorine (C–F) bond activation of the trifluoromethyl group is rarely achieved not only because of the high bond energy but also presumably because of the shielding effect of the lone-pair electrons of the three fluorine atoms.¹ Although defluorinative functionalization of trifluoromethyl-bearing compounds would realize one of the most straightforward approaches to fluorine-containing compounds, harsh reaction conditions have typically been required to cleave C(sp³)–F bonds of trifluoromethyl groups.^{1,2}

Recently, we reported the nickel-mediated [3 + 2] cycloaddition of 2-trifluoromethyl-1-alkenes **1** and alkynes **2** via double C–F bond cleavage of a trifluoromethyl group under mild reaction conditions (Scheme 1a).³ In this reaction, ring-opening of nickelacycle **A**, formed via oxidative cyclization of trifluoromethylalkenes **1** and alkynes **2** with Ni(0), readily proceeded via β -fluorine elimination⁴ to generate alkenylnickel(II) species **B**. Subsequent 5-*endo* insertion and a second β -fluorine elimination afforded 2-fluoro-1,3-cyclopentadienes **3** (Scheme 1, path a). The potential advantage of β -fluorine elimination prompted us to develop a nickel-catalyzed three-component coupling reaction of 2-trifluoromethyl-1-alkenes **1**, alkynes **2**, and metal species, which would proceed via the selective cleavage of one of the C(sp³)–F bonds.^{5,6} We assumed that alkenylnickel(II) fluorides **B**, intermediates toward 2-fluoro-1,3-cyclopentadienes **3**, might be transmetalated, for example, by an appropriate metal hydride (R = H) to afford the corresponding 1,1-difluoro-1,4-dienes **4** along with regeneration of Ni(0) (Scheme 1b, path b).⁷

To prove our hypothesis, we sought a metal hydride reagent suitable for the coupling reaction of α -trifluoromethylstyrene (**1a**) and 4-octyne (**2a**) in the presence of a catalytic amount of Ni(cod)₂ and PCy₃ in toluene at 50 °C (Table 1). In the absence of any hydride sources, fluorocyclopentadiene **3aa** was obtained in 3% yield, as we reported previously (Table 1, entry 1). The use of *i*-PrONa as a hydride source afforded 1,1-difluoro-1,4-diene **4aa** with the *E* configuration as the sole product in 74% yield via cleavage of the C–F bond in the trifluoromethyl group and formation of the C–C and C–H bonds (entry 2).^{7a} When 9-BBN or DIBAL-H was employed, **1a** was decomposed to give a complex mixture because of their electrophilic

Scheme 1. Ni-Catalyzed Defluorinative Coupling of 2-Trifluoromethyl-1-alkenes **1** with Alkynes **2**



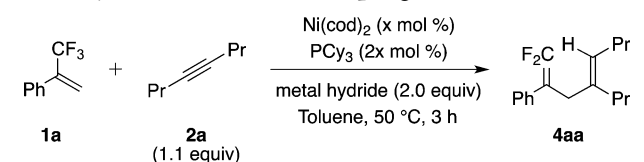
reactivity (entries 3 and 4). Consequently, Et₃SiH was found to be highly effective for improving the product yield to 92% (entry 5).^{7b} Even 5 mol % of the Ni catalyst successfully promoted the coupling reaction to give **4aa** in 93% isolated yield (entry 6).

The scope of suitable trifluoromethylalkene substrates **1** and alkynes **2** was then examined under the reaction conditions

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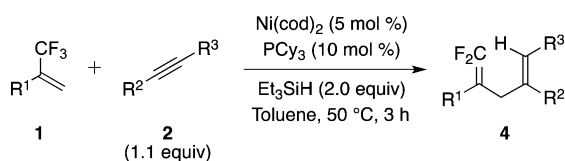
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Table 1. Optimization of the Reaction Conditions for Ni-Catalyzed Defluorinative Coupling of 1a with 2a

entry	x/mol %	metal hydride	4aa/% ^a	3aa/% ^a
1	10	none	0	3
2	10	<i>i</i> -PrONa	74	0
3	10	9-BBN	15	0
4	10	DIBAL-H	0	0
5	10	Et ₃ SiH	92	0
6	5	Et ₃ SiH	93 ^b	0

^aYield was determined by ¹⁹F NMR measurement using PhCF₃ as an internal standard. ^bIsolated yield.

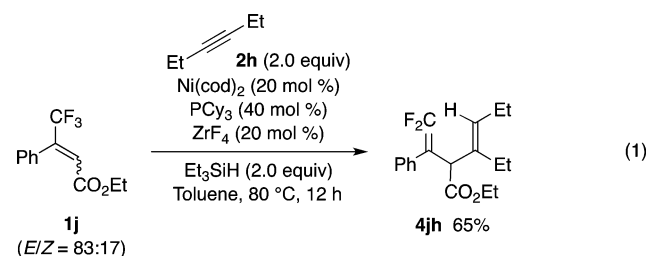
Table 2. Synthesis of 1,1-Difluoro-1,4-dienes 3 via Ni-Catalyzed Defluorinative Coupling of 1 with 2^a

entry	R ¹ (1)	R ² , R ³ (2)	yield/%
1	Ph (1a)	Pr, Pr (2a)	4aa 93
2	C ₆ H ₄ (<i>o</i> -OMe) (1b)	Pr, Pr (2a)	4ba 84
3	C ₆ H ₄ (<i>p</i> -OMe) (1c)	Pr, Pr (2a)	4ca 80
4	C ₆ H ₄ (<i>p</i> -Ac) (1d)	Pr, Pr (2a)	4da 94
5 ^b	C ₆ H ₄ (<i>p</i> -CO ₂ Et) (1e)	Pr, Pr (2a)	4ea 88
6 ^c	C ₆ H ₄ (<i>p</i> -Cl) (1f)	Pr, Pr (2a)	4fa 91
7 ^{d,e}	CH ₂ CH ₂ Ph (1g)	Pr, Pr (2a)	4ga 86
8 ^{d,f}	SiMe ₂ Ph (1h)	Pr, Pr (2a)	4ha 79
9 ^{g,h}	Ph (1a)	Ph, Ph (2b)	4ab 72
10 ⁱ	H (1i)	Ph, Ph (2b)	4ib 77
11 ^{g,j}	Ph (1a)	Pr, Ph (2c)	4ac 99
12 ^{g,j}	Ph (1a)	Pr, C ₆ H ₄ (<i>p</i> -OMe) (2d)	4ad 99
13 ^{g,j}	Ph (1a)	Pr, C ₆ H ₄ (<i>p</i> -CO ₂ Et) (2e)	4ae 65
14 ^j	Ph (1a)	Me, Ph (2f)	4af 91
15 ^k	Ph (1a)	<i>i</i> -Pr, Me (2g)	4ag 88 (95:5) ^l

^aReaction conditions: Ni(cod)₂ (5 mol %), PCy₃ (10 mol %), 1 (0.50 mmol), 2 (0.55 mmol), Et₃SiH (1.0 mmol), toluene (2.5 mL), 50 °C, 3 h. ^b4 h. ^c2 h. ^dNi(cod)₂ (10 mol %), PCy₃ (20 mol %) and ZrF₄ (10 mol %) were used as catalysts. ^e80 °C, 15 h. ^fRT, 2 h. ^gSIMes-HCl (5 mol %) and *t*-BuOK (5 mol %) were used instead of PCy₃. ^hRT, 8 h. ⁱReaction was performed using Ni(cod)₂ (10 mol %), SIMesHCl (10 mol %), *t*-BuOK (10 mol %), 1i (1.0 atm), 2b (0.37 mmol), Et₃SiH (0.74 mmol), toluene (1.9 mL), 80 °C, 10 h. ^jRT, 3 h. ^kNi(cod)₂ (10 mol %), PCy₃ (20 mol %), 4 h. ^lRegioisomer ratio was determined by ¹⁹F NMR.

obtained above (Table 2). The α -trifluoromethylstyrenes 1b and 1c bearing electron-donating methoxy groups provided 1,1-difluoro-1,4-dienes 4ba and 4ca, in good yields (entries 2 and 3, respectively), as did the α -trifluoromethylstyrenes 1d and 1e bearing electron-withdrawing acetyl and ethoxycarbonyl groups (4da and 4ea, entries 4 and 5, respectively). Intriguingly, α -trifluoromethylstyrene 1f with a chlorine substituent, which could undergo oxidative addition to Ni(0), also participated in the reaction without the loss of the C–Cl bond (entry 6). Alkyl- and

silyl-substituted trifluoropropenes 1g and 1h successfully underwent the coupling reaction with 2a in the presence of 10 mol % ZrF₄ as a cocatalyst, leading to high yields of 4ga and 4ha (entries 7 and 8), respectively.⁸ The use of diphenylacetylene (2b) with 1a resulted in the formation of the corresponding coupling product 4ab in 72% yield (entry 9). In the case of coupling between 2b and 3,3,3-trifluoropropene (1i), use of SIMes instead of PCy₃ promoted the reaction to afford 4ib in 77% yield (entry 10). Unsymmetrical 1-phenylpent-1-yne (2c), 1-(4'-methoxyphenyl)pent-1-yne (2d), 1-(4'-ethoxycarbonylphenyl)pent-1-yne (2e), 1-phenylprop-1-yne (2f), and 4-methylpent-2-yne (2g) also participated in this reaction with 1a to afford the corresponding 1,1-difluoro-1,4-dienes 4ac–4ag in good to excellent yields with good to complete regioselectivities (entries 11–15). The obtained regioselectivities were in agreement with literature reports of nickel-catalyzed coupling reactions of alkynes via oxidative cyclization.⁹ Furthermore, the reaction of β -trifluoromethylcinnamate 1j (*E/Z* = 83:17) successfully proceeded with 3-hexyne (2h) to afford the corresponding 1,1-difluoro-1,4-diene 4jh in 65% yield eq 1.

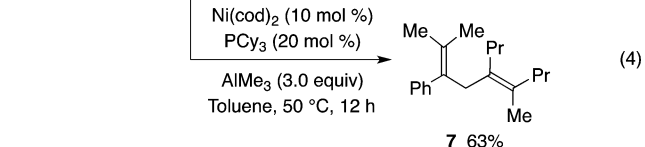
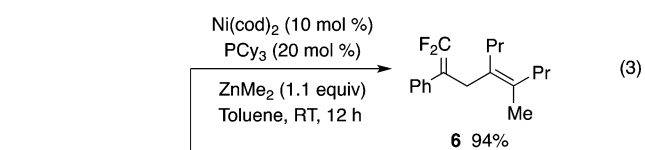
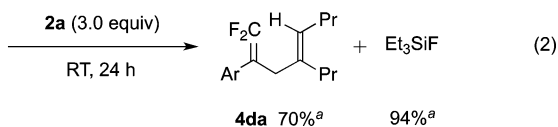
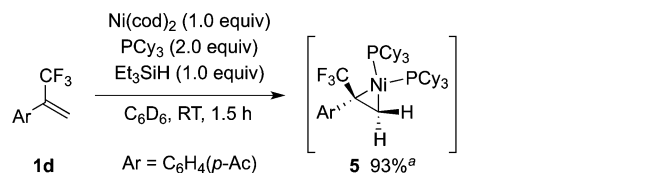
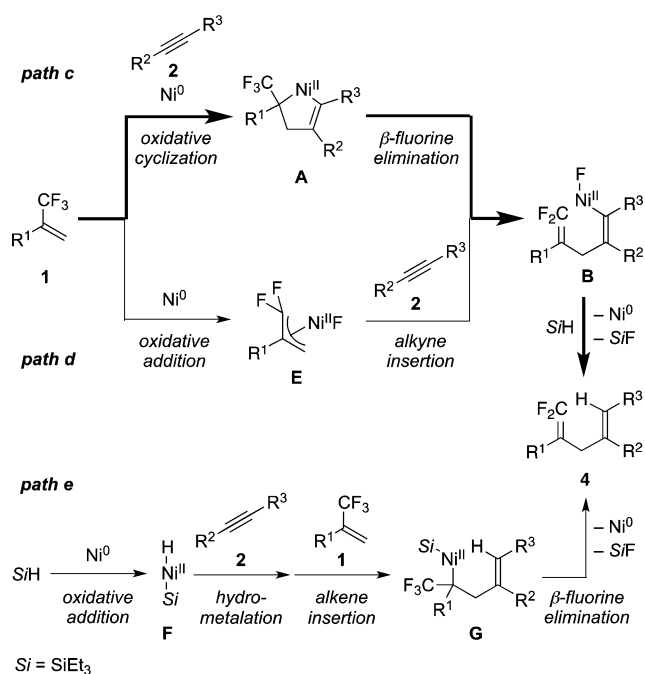


For this reaction, there are three plausible mechanisms that can be induced by different initial steps: (i) oxidative cyclization of 2-trifluoromethyl-1-alkenes 1 and alkynes 2 with Ni(0) (Scheme 2, path c), (ii) oxidative addition of a C–F bond of 1 to Ni(0) (path d),¹⁰ and (iii) oxidative addition of an Si–H bond to Ni(0) (path e).^{11,12} In paths c and d, the common intermediates B are formed via an oxidative cyclization/ β -fluorine elimination or an oxidative addition/insertion sequence. Transmetalation of B with Et₃SiH and subsequent reductive elimination afford 1,1-difluoro-1,4-dienes 4. Conversely, in path e, oxidative addition of Et₃SiH to Ni(0) initially occurs to provide silylnickel hydride F. Subsequent hydrometalation of alkyne and alkene insertion followed by β -fluorine elimination from the alkylnickel complexes G gives 4.

To elucidate the mechanism, the stoichiometric reaction of 2-trifluoromethyl-1-alkene 1d and Et₃SiH with a Ni(0) complex in C₆D₆ at room temperature was performed and monitored by ¹H, ¹⁹F, and ³¹P NMR eq 2. Nickelacyclopropane 5 was obtained as the sole product in 93% yield, and neither allylnickel(II) complex E nor silylnickel hydride F generated by oxidative addition of 1d or Et₃SiH to Ni(0) was observed. Thus, the possibility of path e was lessened. Moreover, treatment of the obtained reaction mixture with alkyne 2a afforded the coupling product 4da in 70% yield from 1d along with 94% yield of Et₃SiF. These results suggest that the C–F bond activation may proceed via an oxidative cyclization/ β -fluorine elimination sequence in this reaction (Scheme 2, path c). Note that the oxidative cyclization process determined the stereochemistry of alkyne-derived alkene moieties of 4 (*syn* addition).

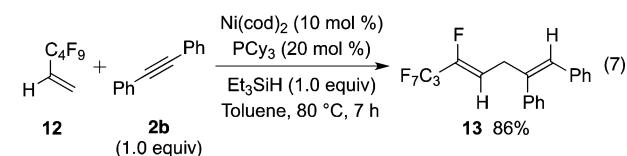
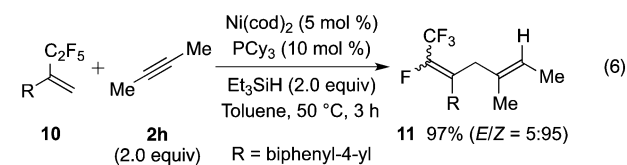
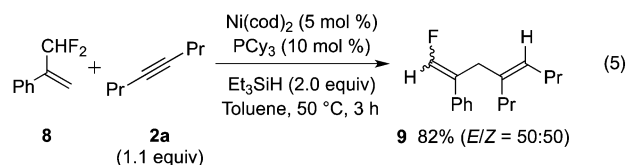
Organozinc and organoaluminum reagents were also employed as a third component in the catalytic coupling reactions eqs 3 and 4.

Scheme 2. Plausible Reaction Mechanisms for the Ni-Catalyzed Defluorinative Coupling Reaction



On treatment with ZnMe₂, 2-trifluoromethyl-1-alkene **1a** and alkyne **2a** underwent the nickel-catalyzed defluorinative coupling to afford methylated 1,1-difluoro-1,4-dienes **6** eq 3. Transmetalation of alkenylnickel(II) fluorides **B** with ZnMe₂ caused introduction of a methyl group into the product. In contrast, the reaction of **1a** and **2a** with AlMe₃ afforded triply methylated 1,4-diene **7** via cleavage of the three C–F bonds eq 4.^{2b,13}

Furthermore, not only trifluoromethylalkenes but also difluoroallylic compounds underwent nickel-catalyzed defluorinative coupling eqs 5–7. The reaction of α -difluoromethylstyrene (**8**)



with alkyne **2a** was promoted by the nickel catalyst in the presence of 2.0 equiv of Et₃SiH. Allylic C–F bond cleavage afforded the corresponding 1-fluoro-1,4-diene **9** in 82% yield (*E/Z* = 50:50, eq 5). Regioselective C–F bond activation of perfluoroalkyl alkenes was also affected by this method eqs 6 and 7. The 2-pentafluoroethyl-1-alkene **10** readily reacted with 2-butyne (**2h**) and Et₃SiH in the presence of the nickel catalyst. Allylic C–F bond cleavage selectively occurred to afford the corresponding trifluoromethylated fluorodiene **11** in 97% yield with high stereoselectivity for the *Z* fluoroalkene moiety (*E/Z* = 5:95, eq 6).¹⁴ Similarly, the reaction of 2-nonafluorobutyl-1-alkene **12** smoothly proceeded to afford heptafluoropropylated fluorodiene **13** in 86% yield with exclusive *Z*-selectivity in the fluoroalkene moiety eq 7.¹⁴

In summary, we have developed a methodology for catalytic C(sp³)–F bond activation of the trifluoromethyl group via β -fluorine elimination from nickelacyclopentenes generated by the stepwise oxidative cyclization of 2-trifluoromethyl-1-alkenes **1** and alkynes **2** with Ni(0). The metal–fluorine bond in the intermediary alkenylnickel(II) fluoride is effectively transformed to a metal–hydrogen (methyl) bond by a hydrosilane (a dialkylzinc), regenerating Ni(0). This reaction provides a regio- and stereoselective method for the synthesis of multisubstituted difluoroalkenes, which have attracted considerable attention as bioisosteres of carbonyl compounds in pharmaceutical science¹⁵ and as building blocks for further transformation into fluorine-containing compounds.^{1,16}

■ ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acscatal.5b01463.

Experimental procedures and characterization data of new compounds (PDF)

Crystallographic data (CIF)

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Notes

The authors declare no competing financial interest.

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- (13) In the presence of 10 mol % of Ni(cod)₂ and 20 mol % of PCy₃, the reaction of compound **6** with 2.0 equiv of AlMe₃ in toluene at 50 °C for 7 h afforded compound **7** in 54% yield, while no reaction occurred without the nickel catalyst. This indicates that the nickel catalyst is required for the substitution of vinylic C–F bonds of the difluoroalkene moiety with AlMe₃.
- (14) The stereoselectivity of the fluoroalkene moiety was determined in the β -fluorine elimination step, which was likely controlled by steric effects. In the reaction of **10**, *syn*- β -fluorine elimination from nickelacycle **A** proceeded to avoid the steric hindrance between the trifluoromethyl group and the biphenyl-4-yl group. Conversely, when **12** was used as the substrate, *syn*- β -fluorine elimination occurred to avoid the steric hindrance between the heptafluoropropyl group and the methylene of nickelacycle **A**.
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